





Ist Floor, Skylark Building, Newal Kishore Road, Hazratgani, Lucknow.

Call: 7080111582, 7080111595

SAMPLE PAPER - 107

Time: 1:15 Hr. Question: 60

PHYSICS

- 01. Electron in hydrogen atom first jumps from third excited state to second excited state and then from second excited to the first excited state. The ratio of the wavelengths $\lambda_1 : \lambda_2$ emitted in the two cases is

- (1) $\frac{7}{5}$ (2) $\frac{27}{20}$ (3) $\frac{27}{5}$ (4) $\frac{20}{7}$
- 02. The temperature dependence of resistance of Cu and undoped Si in the temperature range 300 – 400 K is best described by
 - (1) linear increase for Cu, exponential decrease for Si
 - (2) linear decrease for Cu, linear decrease for Si
 - (3) linear increase for Cu, linear increase for Si
 - (4) linear increase for Cu, exponential increase for Si
- The red light of a wavelength 6400 Å in the air has a 03. wavelength of 4000 Å in glass. If the wavelength of violet light in air is 4400 Å, then the wavelength in the glass is $(\mu_{\text{violet}} = \mu_{\text{red}})$
 - $(1)2570 \,\text{Å}$
- (2) 2750 Å (4) 2510 Å
- $(3)\,1600\,\text{Å}$
- 04. A body of mass m slides down an incline and reaches the bottom with a velocity v. If the same mass were in the form of a ring which rolls down this incline, the velocity of the ring at bottom would have been:
 - (1) v

- (2) $\sqrt{2}$ v (3) $\frac{1}{\sqrt{2}}$ v (4) $\sqrt{\frac{2}{5}}$ v
- 05. A thin horizontal circular disc is rotating about a vertical axis passing through its centre. An insect is at rest at a point near the rim of the disc. The insect now moves along a chord of the disc to reach its other end near rim. During the journey of the insect, the angular speed of the disc.
 - (1) remains unchanged
 - (2) first decreases and then increases
 - (3) first increases and then decreases
 - (4) continuously increases

- 06. A transverse harmonic wave on a string is described by $y(x,t) = 3.0 \sin (36t + 0.018x + \pi/4)$ where x and y are in cm and t is in s. The positive direction of x is from left to
 - (I) The wave is travelling from right to left
 - (II) The speed of the wave is 20 m/s
 - (III) Frequency of the wave is 5.7 Hz.
 - (IV) The least distance between two successive crests in the wave is 2.5 cm.

The correct options are:

- (1) I, II only
- (2) I, III only
- (3) I, II, III only
- (4) all
- 07. Two conducting plates A and B are placed parallel to each other at a small distance between them. Plate A is given a charge q_1 and plate B is given a charge q_2 . Then
 - (1) the outer surfaces of A and B get no charge
 - (2) the inner surfaces of A and B get all the charge
 - (3) the inner surfaces of A and B get equal and opposite

charge of magnitude $\frac{q_1 - q_2}{2}$

(4) the outer surfaces of A and B get charge of the same

polarity and of the magnitude $\frac{q_1 - q_2}{2}$

- 08. In vibration magnetometer if length of a thin bar magnetic is shortened to $\frac{1}{4}$ times keeping its cross-sectional area constant, then time period will become
 - $(1) \frac{1}{2}$ times
- (3) 2 times
- (4) $\frac{1}{4}$ times
- 09. The inductance of a coil is L = 10 H and resistance $R = 5\Omega$. If applied voltage of battery is 10 V and it switches OFF in 1ms, then find the value of induced emf of inductor $(1) 2 \times 10^4 \text{ V}$
 - $(2) 1.2 \times 10^4 \text{ V}$
 - $(3) 2 \times 10^{-4} \text{ V}$
 - (4) None of these

10. The moment of inertia of a uniform disc about an axis perpendicular to plane and through centre is $\frac{1}{2}$ MR²(M

> = mass, R = radius of disc). If the disk is rolling on its edge without slipping on a straight line path, then the ratio of rotational kinetic energy to translational one is:

- (1)1

- (2) $\frac{1}{2}$ (3) $\frac{1}{4}$ (4) $\frac{1}{8}$
- 11. A body moves with speed V₁ for distance L and then with speed V₂ for distance 2L. The average speed for the

 - (1) $\frac{V_1 + V_2}{2}$ (2) $\frac{3V_1V_2}{V_1 + 2V_2}$
 - (3) $\frac{3V_1V_2}{2V_1 + V_2}$ (4) $\frac{3V_1V_2}{V_1 + V_2}$
- 12. A stone of mass 0.3 kg attached to a 1.5 m long string is whirled around in a horizontal circle on a frictionless table at a speed of 6 ms⁻¹. The tension in the string is
 - (1) 10 N
- (2)20N
- (3)7.2N
- (4) None of these
- 13. A coin placed on a rotating table just slips if it is placed at a distance 4r from the centre. If we double the angular velocity of the table, then the coin will just slip when it is away from the centre at a distance equal to
 - (1)4r
- (2) 2r
- (3) r
- 14. An elevator weighing 6000 kg is pulled upwards by a cable with an acceleration of 5 ms⁻². Taking g to be 10 ms⁻², the tension in the cable is
 - (1)6000 N
- (2) 9000 N
- (3)60000 N
- (4)90000 N
- A pendulum of length 1 m is released from $\theta = 60^{\circ}$. The 15. rate of change of speed of the bob, at $\theta = 30^{\circ}$, is (g=10 ms^{-2})
 - $(1) 10 \,\mathrm{m}\,\mathrm{s}^{-2}$
- $(3) 5 \text{ m s}^{-2}$
- (4) $5\sqrt{3}$ m s⁻²

CHEMISTRY

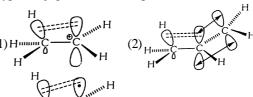
- Atomic radii of Zr (160 pm) and Hf (159 pm) is a 16. consequence of-
 - (1) Transition contraction
 - (2) Inert pair effect
 - (3) poor shielding of 4f electrons
 - (4) Actinoide contraction
- 17. Bond order increases in which of the cases given below?
 - $(1) CO \rightarrow CO^+$
- (2) $N_2 \rightarrow N_2^-$
- $(3) O_2 \rightarrow O_2^{-2}$
- $(4) H_2 \rightarrow H_2^-$

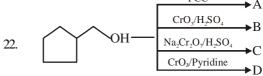
- Which statement is not true about N_2O_5 ? 18.
 - (1) It is anhydride of HNO₃
 - (2) In solid state it exists as NO₂+NO₃
 - (3) It is structurally similar to P_4O_{10}
 - (4) It can be prepared by heating HNO₃ over P₂O₅
- 19. Select the correct statement(s) given below for amorphous form of carbon:
 - (I) Carbon-black, coke and charcoal are impure amorphous allotropes of carbon.
 - (II) Carbon-black is obtained by burning hydrocarbons or high carbon-content oils in limited supply of air.
 - (III) Charcoal and coke are obtained by heating wood or coal respectively at high temperature in the absence of air.
 - (1) (I) only
- (2) (I) and (II)
- (3) (I) and (III)
- (4) All (I), (II) and (III)
- 20. Correct order of stability of the following resonating structure is

(a)
$$CH_2 = CH - C - H \leftrightarrow$$

$$(b) \overset{\oplus}{C} \overset{|}{H}_2 - CH = \overset{|}{C} - H$$

- (1) a > b > c
- (2) a > c > b
- (3) b > a > c
- (4) c > b > a
- Hyperconjugation can be represented as:





Identify A, B, C and D

(1)
$$A = D = CHO$$
; $B = C = COOH$
(2) $A = D = COOH$; $B = C = CHO$

(4)
$$A = CHO$$
 CHO
 CHO
 CHO
 CHO
 CHO
 CHO
 CHO

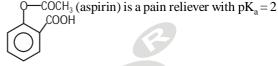
- 23. An organic compound contains 69.77% carbon, 11.63% hydrogen and rest of oxygen. The molecular mass of the compound is 86. It does not reduce tollen's reagent but forms an addition compound with sodium hydrogen sulphite and gives positive iodoform test. On vigrous oxidation it gives ethanoic acid and propanoic acid. The possible structure of the compound is:
 - (1) $CH_3 CH CH_2 CH = CH_2$

 - (4) $CH_3 C CH_2 CH_2 CH_3$
- 24. Aniline on acetylation followed by treatment with nitrating mixture gives a product which on hydrolysis gives compound 'X'. X is
- NHCOCH₃
- HCOCH₂
- \rightarrow CH₃-CH-CH₂-NH₂ Find out the 25.

Reduction Product of 'A'

- (1) Iso Amyl amine
- (2) Active amyl amine
- (3) Active Iso amine
- (4) Tertiary amyl amine
- 26. The values of observed and calculated molecular weights of silver nitrate are 92.64 and 170, respectively. The degree of dissociation of silver nitrate is
- (2)83.5%
- (3) 46.7%
- (4) 60.23%
- 27. Two moles of zinc is dissolved in HCl at 25°C. The work done in open vessel is
 - (1)-2.477 kJ
- (2) 4.955 kJ
- $(4) 0.0489 \, kJ$
- (4) none

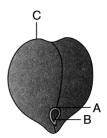
- 28. $CH_3 - CO - CH_{3(g)} \rightleftharpoons CH_3 - CH_{3(g)} + CO_{(g)}$ Initial pressure of CH₃COCH₃ is 100 mm. When equilibrium is set up, mole fraction of CO(g) is 1/3. Hence, K_p is
 - $(1)100 \, \text{mm}$
- $(2)50 \, \text{mm}$
- $(3) 25 \, \text{mm}$
- (4) 150 mm
- 29. 2 mol PCl₅, 1 mol Cl₂ and 1 mol PCl₃ are taken in 1 L flask. When equilibrium is set up, PCl₅ is found to 50% dissociated into the products. Hence, K_c is
 - $(1) 4 \text{ mol } L^{-1}$
- $(2) 1.5 \, \text{mol L}^{-1}$
- (3) $1 \text{ mol } L^{-1}$
- $(4) 0.17 \, \text{mol L}^{-1}$
- 30.



- Two tablets each containing 0.09 g of aspirin are dissolved in 100 mL solution. pH will be (molar mass of aspirin = 180)
- (1)0.5
- (2)1.0
- (3)0.0
- (4)2.0

BOTANY

- 31. In a ring girdled plant
 - (1) The shoot and root die together
 - (2) Neither root nor shoot will die
 - (3) The shoot dies first
 - (4) The root dies first
- 32. Which product of glycolysis is consumed in alcoholic fermentation?
 - (1) NADH + H^+
- (2) ATP
- (3) ATP and NADH + H⁺ (4) CO₂
- 33. Abscisic acid is called stress hormone as it
 - (1) Induces flowering
 - (2) Breaks seed dormancy
 - (3) Promotes leaf fall
 - (4) Promotes stomatal closure
- 34. Identify A to C in dicotyledonous seed.

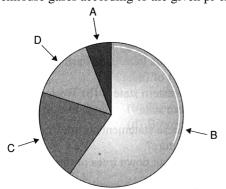


- (1) A-Hilum, B-Micropyle, C-Seed coat
- (2) A-Seed coat, B-Hilum, C-Micropyle
- (3) A-Micropyle, B-Seed coat, C-Hilum
- (4) A-Micropyle, B-Hilum, D-Seed coat

- 35. Which one of the following organisms is correctly matched with its three characteristics?
 - (1) Tomato: Twisted aestivation, Axile placentation, Berry
 - (2) Onion: Bulb, Valvate aestivation, Axile placentation
 - (3) Maize: C₃ pathway, closed vascular bundles,
 - (4) Pea: C₃ pathway, Endospermic seed, Vaxillary aestivation
- Choose the correct sequence representing the ploidy of 36. Nucellus; Megaspore mother cell; Megaspore; Egg cell; Zygote; A polar nucleus of embryo sac; Secondary nucleus and primary endosperm nucleus.
 - (1) n; 2n; 2n; n; 2n; n; 2n; and 3n
 - (2) 2n; 2n; n; 2n; n; 2n; 3n; and 3n
 - (3) 2n; 2n; n; n; 2n; n; 2n; and 3n
 - (4) 2n; n; n; 2n; 3n; 2n; n; and 3n
- 37. The name of Norman Borlaug is associated with:
 - (1) White Revolution
- (2) Green Revolution
- (3) Yellow Revolution
- (4) Blue Revolution
- 38. Fill in the blanks A, B, C, and D respectively

Species A	Species B	None of interaction
+	+	Mutualism
_	_	A
+	_	B
+	-	Parasitism
+	0	C
_	0	D

- (1) A-Commensalism, B-Predation, C-Amensalism, D-Competition
- (2) A-Predation, B-Parasitism, C-Commensalism, D-Amensalism
- (3) A-Competition, B-Predation, C-Commensalism, D-Amensalism
- (4) A-Competition, B-Predation, C-Amensalism, D-Commensalism
- 39. Out of the total cost of various ecosystem services, the soil account for ______%, recreation and nutrient cycling are less than % each and climate regulation and _____% each. habitat for wildlife is about _
 - (1)50, 6, 10
- (2)50, 10, 6
- (3) 50, 30, 20
- (4)20, 30, 50
- 40. What is the relative contribution (A, B, C, D) of various greenhouse gases according to the given pi-chart? A



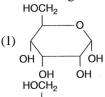
- (1) A–N₂O (6%), B–Carbon dioxide (60%), C–Methane (20%), D-CFCs (14%).
- (2) A-Methane (20%), B-CFCs (14%), C-N₂O (6%), D-Carbon dioxide (60%).
- (3) A–Carbon dioxide (6%), B–N₂O (6%), C–Methane (20%), D-CFCs (14%).
- (4) A–CFCs (14%), B–Carbon dioxide (60%), C–Methane (20%), D-N₂O (6%).
- 41. The process of guttation takes place:
 - (1) when the root pressure is high and the rate of transpiration is low
 - (2) when the root pressure is low and the rate of transpiration is high
 - (3) when the root pressure equals the rate of transpiration
 - (4) when the root pressure as well as rate of transpiration are high
- 42. Match the followings and choose the correct option:
 - (A) Leaves
- (i) Anti-transpirant
- (B) Seeds
- (ii) Transpiration
- (C) Roots
- (iii) Negative osmotic
- potential
- (D) Aspirin
- (iv) Imbibition (v) Absorption
- (E) Plasmolyzed cell
- (1) (A)-(iii), (B)-(iv), (C)-(i), (D)-(ii), (E)-(v)(2) (A)-(ii), (B)-(iv), (C)-(v), (D)-(i), (E)-(iii)
- (3)(A)-(iii),(B)-(ii),(C)-(iv),(D)-(i),(E)-(v)
- (4)(A)-(iii),(B)-(ii),(C)-(i),(D)-(iv),(E)-(v)
- 43. Match the element with its associated functions/roles and choose the correct option among given below:
 - (1) Boron
- (i) Splitting of H₂O to
- liberate O₂ during
- photosynthesis
- (2) Manganese
- (ii) Needed for synthesis
- of auxins
- (3) Molybdenum
- (iii) Component of
- nitrogenase
- (4) Zinc
- (iv) Pollen germination
- (E) Iron
- (v) Component of
- ferredoxin
- (1)(1)-(i), (2)-(ii), (3)-(iii), (4)-(iv), (E)-(v)
- (2)(1)-(iv), (2)-(i), (3)-(iii), (4)-(ii), (E)-(v)
- (3)(1)-(iii), (2)-(ii), (3)-(iv), (4)-(v), (E)-(i)
- (4)(1)-(ii), (2)-(iii), (3)-(v), (4)-(i), (E)-(iv)
- 44. Which light range is least effective in photosynthesis?
 - (1) Blue
- (2) Green
- (3) Red
- (4) Motet
- 45. The reaction that is responsible for the primary fixation CO₂ is catalysed by
 - (1) RuBP carboxylase
 - (2) PEP carboxylase
 - (3) RuBP carboxylase and PEP carboxylase
 - (4) PGA synthase

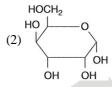
ZOOLOGY

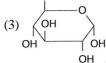
- 46. The intestine and stomach in mammals are lined by
 - (1) Cuboidal epithelium (2) Columnar epithelium
 - (3) Squamous epithelium (4) Stratified epithelium
- 47. Osteomalacia is due to deficiency of :
 - (1) vitamin-A
- (2) vitamin-C
- (3) vitamin-E
- (4) vitamin-D
- 48. The below diagram shows _____ symmetry, which is also found in the following group of organisms.



- (1) Radial, Adamsia, Asterias, Aplysia
- (2) Bilateral, Salpa, Hyla, Calotes
- (3) Bilateral Taenia, Ctenoplana, Antedon
- (4) Radial, Doliolum, Gorgonia, Sycon
- 49. Which of the following represents the correct Haworth structure of glucose?







- (4) All of these
- 50. Match the columns.

	Column–I (Category)		Column-II (Secon dary metabolites)
(A)	Pigments	(1)	Concanavalin A
(B)	Terpenoids	(2)	Monoterpenes, Diterpenes
(C)	Alkaloids	(3)	Morphine, Codeine
(D)	Lectins	(4)	Carotenoids, Anthocyanin
(E)	Toxins	(5)	Abrin, Ricin
(F)	Drugs	(6)	Vinblastine, Curcumin

- (1) A-1, B-3, C-2, D-4, E-5, F-6
- (2) A-4, B-2, C-3, D-1, E-5, F-6
- (3) A-6, B-3, C-5, D-2, E-4, F-1
- (4) A-1, B-2, C-3, D-4, E-5, F-6
- 51. The total thickness of respiratory diffusion membrane is
 - (1) Less than fm
 - (2) Less than micrometre
 - (3) Much less than mm
 - (4) Less than nm
- 52. Which of the following statement is correct about pseudopodia?
 - (1) Formed by streaming of protoplasm
 - (2) Formed in amoeba and neutrophil
 - (3) Both (1) and (2)
- (4) None of these

- 53. Foetal limbs and digits are formed in the _____ month of embryonic development.
 - $(1)4^{th}$
- $(2) 2^{nd}$
- $(3) 3^{rd}$
- $(4)7^{th}$
- 54. Which of these options is correct with regards to statements X and Y?

Statement X: Some STDs do not show symptoms in females.

Statement Y: Some STDs in females may remain undetected for long time.

- (1) Statement 'X' and 'Y' are correct and 'X' is the correct explanation for 'Y'.
- (2) Only statement 'X' is correct.
- (3) Only statement 'Y' is correct.
- (4) Statement 'X' and 'Y' are correct.
- 55. The Bt-toxin is not toxic to human beings because
 - (1) The pro Bt-toxin inactivation requires above human body temperature
 - (2) The Bt-toxin recognizes only insect specific target
 - (3) The Bt-toxin formation from pro Bt-toxin requires pH lower than that present in the human stomach
 - (4) Conversion of pro Bt-toxin to Bt-toxin takes place only in highly alkaline conditions
- 56. The major inorganic component of vertebrate bone is
 - (1) sodium chloride
- (2) calcium carbonate

(4) magnesium phosphate

- (3) calcium phosphate (4) 57. Vitamin K is required for
 - Vitamin K is required for (1) synthesis of prothrombin
 - (2) formation of thromboplastin
 - (3) conversion of fibrinogen to fibrin
 - (4) conversion of prothrombin to thrombin
- 58. Blood does not clot inside the vessels because
 - 1. Constant speedy flow of blood prevents accumulation of fibrin threads, if at all formed
 - 2. Absence of heparin, an anticoagulant having antithrombin activity, in blood
 - 3. Smoothness of the endothelial lining prevents rupture by platelets and formation of thromboplastin from them.
 - 4. Presence of monomolecular layer of negatively charged protein adsorbed to the inner surface of endothelium, which attracts the clotting factors
 - (1) 1 and 2 are correct
- (2) 2 and 4 are correct
- (3) 1 and 3 are correct
- (4) 1,2 and 3 are correct
- 59. Wharton's duct is associated with
 - (1) Sublingual salivary gland
 - (2) Parotid salivary gland
 - (3) Submaxillary salivary gland
 - (4) Brunner's gland
- 60. A certain person is eating boiled potato. One of the food components in it is
 - (1) Lactose which is indigestible
 - (2) Starch which does not get digested
 - (3) DNA which gets digested by pancreatic DNase
 - (4) Cellulose which is digested by cellulase